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# Water quality index for assessment of drinking groundwater purpose case study: area surrounding Ismailia Canal, Egypt

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## Abstract

The dramatic increase of different human activities around and along Ismailia Canal threatens the groundwater system. The assessment of groundwater suitability for drinking purpose is needed for groundwater sustainability as a main second source for drinking. The Water Quality Index (WQI) is an approach to identify and assess the drinking groundwater quality suitability.

The analyses are based on Pearson correlation to build the relationship matrix between 20 variables (electrical conductivity (Ec), pH, total dissolved solids (TDS), sodium (Na), potassium (K), calcium (Ca), magnesium (Mg), chloride (Cl), carbonate (CO<sub>3</sub>), sulphate (SO<sub>4</sub>), bicarbonate (HCO<sub>3</sub>), iron (Fe), manganese (Mn), zinc (Zn), copper (Cu), lead (Pb), cobalt (Co), chromium (Cr), cadmium (Cd), and aluminium (Al)). Very strong correlation is found at [Ec with Na, SO<sub>4</sub>] and [Mg with Cl]; strong correlation is found at [TDS with Na, Cl], [Na with Cl, SO<sub>4</sub>], [K with SO<sub>4</sub>], [Mg with SO<sub>4</sub>] and [Cl with SO<sub>4</sub>], [Fe with Al], [Pb with Al]. The water type is Na–Cl in the southern area due to salinity of the Miocene aquifer and Mg–HCO<sub>3</sub> water type in the northern area due to seepage from Ismailia Canal and excess of irrigation water.

The WQI classification for drinking water quality is assigned with excellent and good groundwater classes between km 10 to km 60, km 80 to km 95 and the adjacent areas around Ismailia Canal. While the rest of WQI classification for drinking water quality is assigned with poor, very poor, undesirable and unfit limits which are assigned between km 67 to km 73 and from km 95 to km 128 along Ismailia Canal.

**Keywords:** Groundwater, Ismailia Canal, Suitability, Water Quality Index

## Introduction

Nowadays, groundwater has become an important source of water in Egypt. Water crises and quality are serious concerns in a lot of countries, particularly in arid and semi-arid regions where water scarcity is widespread, and water quality assessment has received minimal attention [3, 9]. So, it is important to assess the quality of water to be used, especially for drinking purposes.

Poor hydrogeological conditions have been encountered causing adverse impacts on threatening the adjacent groundwater aquifer under the Ismailia Canal. The groundwater

quality degradation is due to rapid urban development, industrialization, and unwise water use of agricultural water, either groundwater or surface water.

As groundwater quality is affected by several factors, an appropriate study of groundwater aquifers characteristics is an essential step to state a supportable utilization of groundwater resources for future development and requirements [11, 12]. It is important that hydrogeochemical information is obtained for the region to help improving the groundwater management practices (sustainability and protection from deterioration) [17].

Many researchers have paid great attention to groundwater studies. In the current study area, the hydrogeology and physio-hydrochemistry of groundwater in the current study area had been previously discussed by El Fayoumy [15] and classified the water to NaCl type; Khalil et al. [27] stated that water had high concentration of Na, Ca, Mg, and K. Geriesh et al. [21] detected and monitored a waterlogging problem at the Wadi El Tumilate basin, which increased salinity in the area. Singh [34] studied the problem of salinization on crop yield. Awad et al. [7] revealed that the groundwater salinity ranges between 303 ppm and 16,638 ppm, increasing northward in the area.

Various statistical concepts were used to understand the water quality parameters [24, 28, 35].

Armanuos et al. [4] studied the groundwater quality using WQI in the Western Nile Delta, Egypt. They had generated the spatial distribution map of different parameters of water quality. The results of the computed WQI showed that 45.37% and 66.66% of groundwater wells falls into good categories according to WHO and Egypt standards respectively.

Eltarabily et al. [19] investigate the hydrochemical characteristics of the groundwater at El-Khanka in the eastern Nile Delta to discuss the possibility of groundwater use for agricultural purposes. They used Pearson correlation to deduce the relationship between 13 chemical variables used in their analysis. They concluded that the groundwater is suitable for irrigation use in El-Qalubia Governorate.

The basic goal of WQI is to convert and integrate large numbers of complicated data-sets of the physio-hydrochemistry elements with the hydrogeological parameters (which have sensitive effect on the groundwater system) into quantitative and qualitative water quality data, thus contributing to a better understanding and enhancing the evaluation of water quality [38]. The WQI is calculated by performing a series of computations to convert several values from physicochemical element data into a single value which reflects the water quality level's validity for drinking [16].

Based on the physicochemical properties of the groundwater, it should be appraised for various uses. One can determine whether groundwater is suitable for use or unsafe based on the maximum allowable concentration, which can be local or international. The type of the material surrounding the groundwater or dissolving from the aquifer matrix is usually reflected in the physicochemical parameters of the groundwater. These metrics are critical in determining groundwater quality and are regarded as a useful tool for determining groundwater chemistry and primary control mechanisms [18].

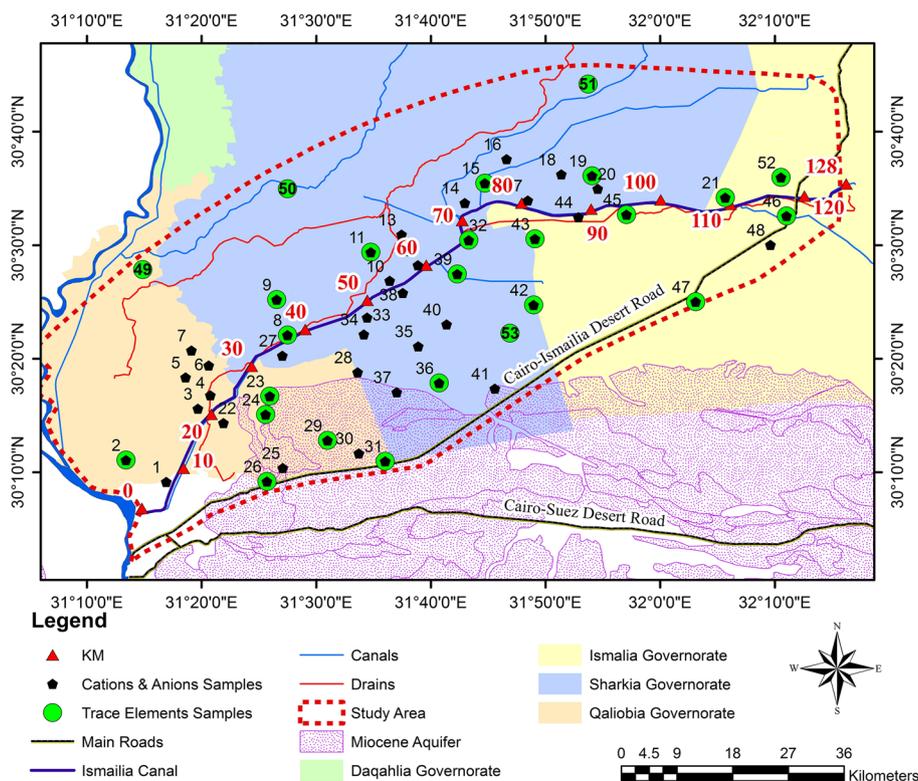
The objective of this research is to assess suitability of groundwater quality of the study area around Ismailia Canal for drinking purpose and generating WQI map to help decision-makers and local authorities to use the created WQI map for

groundwater in order to avoid the contamination of groundwater and to facilitate in selection safely future development areas around Ismailia Canal.

**Description of study area**

The study area lies between latitudes 30° 00' and 31° 00' North and longitude 31° 00' and 32° 30' East. It is bounded by the Nile River in the west, in the east there is the Suez Canal, in the south, there is the Cairo-Ismailia Desert road, and in the north, there are Sharqia and Ismailia Governorates as shown in Fig. 1. Ismailia Canal passes through the study area. It is considered as the main water resource for the whole Eastern Nile Delta and its fringes. Its intake is driven from the Nile River at Shoubra El Kheima, and its outlet at the Suez Canal. At the intake of the canal, there are large industrial areas, which include the activities of the north Cairo power plant, Amyria drinking water plant, petroleum companies, Abu Zabaal fertilizer and chemical company, and Egyptian company of Alum. Ismailia Canal has many sources of pollution, which potentially affects and deteriorates the water quality of the canal [22].

The topography plays an important role in the direction of groundwater. The ground level in the study area is characterized by a small slope northern Ismailia Canal. It drops gently from around 18 m in the south close to El-Qanater El-Khairia to 2 amsl northward. While southern Ismailia Canal, it is characterized by moderate to high slope. The topography rises from 10 m to more than 200 m in the south direction.



**Fig. 1** Map of the study area and location of groundwater wells

### ***Geology and hydrogeology***

The sequence of deposits rocks of wells was investigated through the study of hydrogeological cross-section A-A' and B-B' located in Fig. 2a, b [32]. Section B-B' shows that the study area represents two main aquifers that can be distinguished into the Oligocene aquifer (southern portion of the study area) and the Quaternary aquifer (northern portion of the study area). The Oligocene aquifer dominates the area of Cairo-Suez aquifer foothills. The Quaternary occupies the majority of the Eastern Nile Delta. It consists of Pleistocene sand and gravel. It is overlain by Holocene clay. The aquifer is semi-confined (old flood plain) and is phreatic at fringes areas in the southern portion of eastern Nile Delta fringes. The Quaternary aquifer thickness varies from 300 m (northern of the study area) to 0 at the boundary of the Miocene aquifer (south of the study area). The hydraulic conductivity ranges from 60 m/day to 100 m/day [8]. The transmissivity varies between 10,000 and 20,000 m<sup>2</sup>/day.

### ***Groundwater recharge and discharge***

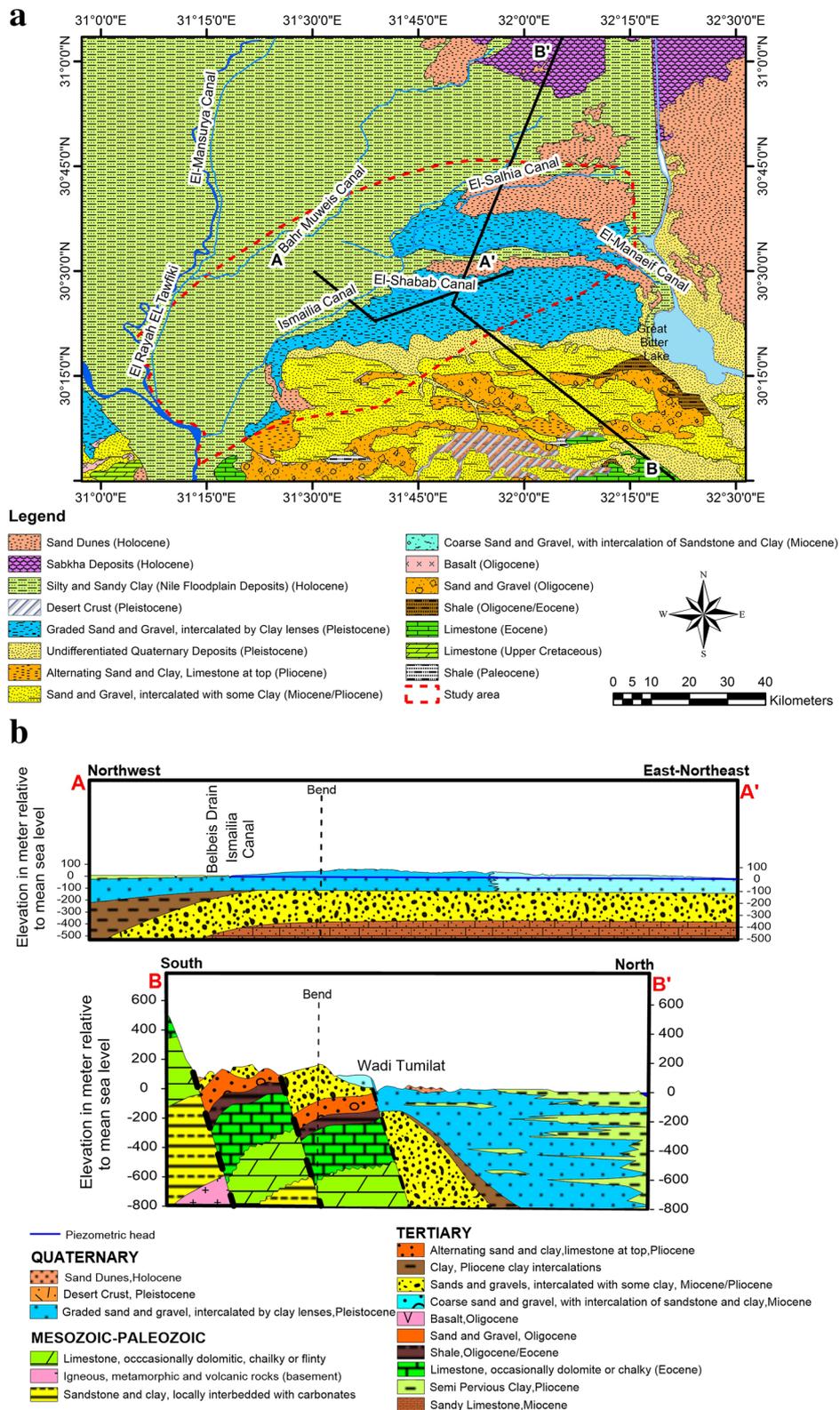
The main source of recharge into the aquifer under the study area is the excess drainage surplus (0.5–1.1 mm/day) [29], in addition to the seepage from irrigation system including Damietta branch and Ismailia Canal.

### ***Groundwater and its movements***

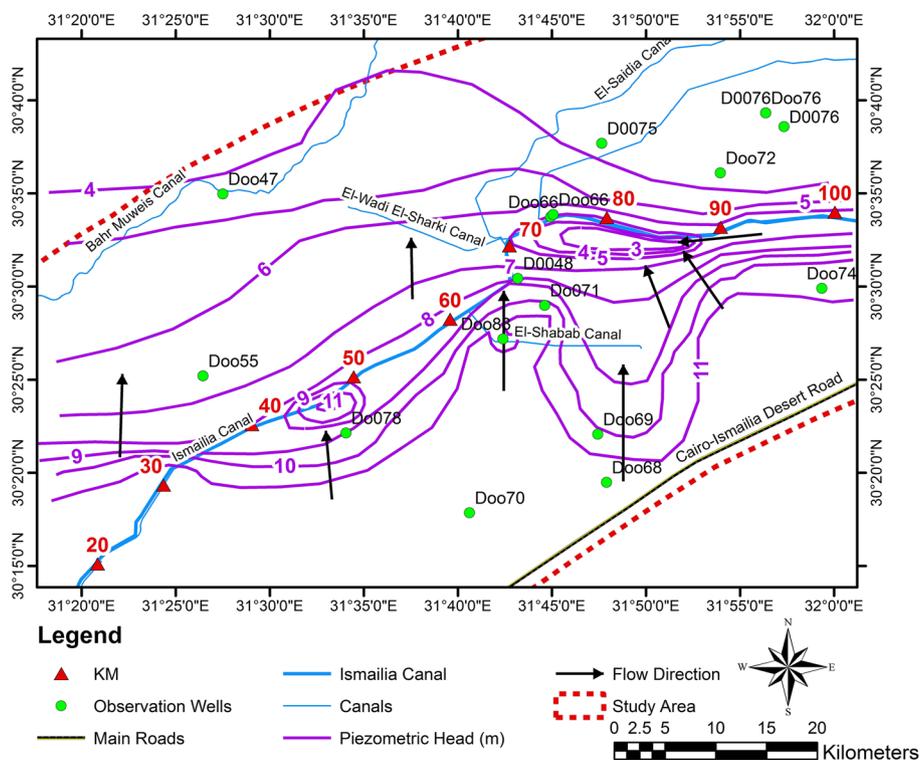
In the current research, it was possible to attempt drawing sub-local contour maps for groundwater level with its movement as shown in Fig. 3. Figure 3 shows the main direction of groundwater flow from south to north. The groundwater levels vary between 5 m and 13 m (above mean sea level). The sensitive areas are affected by (1) the excess drainage surplus from the surface water reclaimed areas which located at low lying areas; (2) the seepage from the Ismailia Canal bed due to the interaction between it and the adjacent groundwater system, and (3) misuse of the irrigation water of the new communities and other issues. Accordingly, a secondary movement was established in a radial direction that is encountered as a source point at the low-lying area (Mullak, Shabab, and Manaief). Groundwater movement acts as a sink at lower groundwater areas (the northern areas of Ismailia Canal located between km 80 to km 90) due to the excessive groundwater extraction. The groundwater level reaches 2 m (AMSL). The groundwater levels range between + 15 m (AMSL) (southern portion of Ismailia Canal and study area near the boundary between the quaternary and Miocene aquifers).

### **Methods**

The assessment of groundwater suitability for drinking purposes is needed and become imperative based on (1) the integration between the effective environmental hydrogeological factors (the selected 9 trace elements Fe, Mn, Zn, Cu, Pb, Co, Cr, Cd, Al) and 11 physio-chemical parameters (major elements of the anions and cations pH, EC, TDS, Na, K, Ca, Mg, Cl, CO<sub>3</sub>, SO<sub>4</sub>, HCO<sub>3</sub>); (2) evaluation of WQI for drinking water according to WHO [36] and drinking Egyptian standards limit [14]; (3) GIS is



**Fig.2** a Geology map of the study area. b Hydrogeological cross-section of the aquifer system (A-A') and geological cross-section for East of Delta (B-B')



**Fig. 3** Groundwater flow direction map in the study area (2019)

used as a very helpful tool for mapping the thematic maps to allocate the spatial distribution for some of hydrochemical parameters with reference standards.

The groundwater quality for drinking water suitability is assessed by collecting 53 water samples from an observation well network covering the area of study, as seen in Fig. 1. The samples were collected after 10 min of pumping and stored in properly washed 2 L of polyethylene bottles in iceboxes until the analyses were finished. The samples for trace elements were acidified with nitric acid to prevent the precipitation of trace elements. They were analyzed by the standard method in the Central Lab of Quality Monitoring according to American Public Health Association [2].

The water quality index is used as it provides a single number (a grade) that expresses overall water quality at a certain location based on several water quality parameters. It is calculated from different water parameters to evaluate the water quality in the area and its potential for drinking purposes [13, 25, 31, 33]. Horton [23] has first used the concept of WQI, which was further developed by many scholars.

The first step of the factor analysis is applying the correlation matrix to measure the degree of the relationship and strength between linearly chemical parameters, using “Pearson correlation matrix” through an excel sheet. The analyses are mainly based on the data from 53 wells for physio-chemical parameters for the major elements and trace elements. Accordingly, it classified the index of correlation into three classes: 95 to 99.9% (very strong correlation); 85 to 94.9% (strong correlation), 70 to 84.9% (moderately), < 70% (weak or negative).

Equation (1) [4] is used to calculate WQI for the effective 20 selected parameters of groundwater quality.

$$WQI = \sum Q_i \times W_i \quad (1)$$

In which  $Q_i$  is the  $i$ th quality rating and is given by equation (2) [4],  $W_i$  is the  $i$ th relative weight of the parameter  $i$  and is given by Eq. (3) [4].

$$Q_i = (C_i/S_i) \times 100 \quad (2)$$

Where  $C_i$  is the  $i$ th concentration of water quality parameter and  $S_i$  is the  $i$ th drinking water quality standard according to the guidelines of WHO [36] and Egypt drinking water standards [14] in milligram per liter.

$$W_i = w_i / \sum_{i=0}^n w_i \quad (3)$$

Where  $W_i$  is the relative weight,  $w_i$  is the weight of  $i$ th parameter and  $n$  is the number of chemical parameters. The weight of each parameter was assigned ( $w_i$ ) according to their relative importance relevant to the water quality as shown in Table 2, which were figured out from the matrix correlation (Pearson correlation, Table 1). Accordingly, it was possible assigning the index for weight ( $w_i$ ). Max weight 5 was assigned to very strong effective parameter for EC, K, Na, Mg, and Cl; weight 4 was assigned to a strong effective parameter as TDS,  $SO_4$ ; 3 for a moderate effective parameter as Ca; and weight 2 was assigned to a weak effective parameter like pH,  $HCO_3$ ,  $CO_3$ , Fe, Cr, Cu, Co, Cd, Pb, Zn, Mn, and Al. Equation (2) was calculated based on the concentration of the collected samples from representative 53 wells and guidelines of WHO [36] and Egypt drinking water standards [14] in milligram per liter. This led to calculation of the relative weight for the weight ( $W_i$ ) by equation (3) of the selected 20 elements (see Table 2). Finally, Eq. (1) is the summation of WQI both the physio-chemical and environmental parameters for each well eventually.

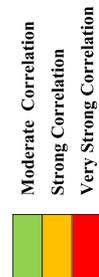
The spatial analysis module GIS software was integrated to generate a map that includes information relating to water quality and its distribution over the study area.

## Results and discussion

The basic statistics of groundwater chemistry and permissible limits WHO were presented in Table 3. It summarized the minimum, maximum, average, med. for all selected 20 parameters and well percentage relevant to the permissible limits for each one; the pH values of groundwater samples ranged from 7.1 to 8.5 with an average value of 7.78 which indicated that the groundwater was alkaline. While TDS ranged from 263 to 5765 mg/l with an average value of 1276 mg/l. Sodium represented the dominant cation in the analyzed groundwater samples as it varied between 31 and 1242 mg/l, with an average value of 270 mg/l. Moreover, sulfate was the most dominant anion which had a broad range (between 12 and 1108 mg/l), with an average value of 184 mg/l. This high sulfate concentration was due to the seepage from excess irrigation water and the dissolution processes of sulfate minerals of soil composition which are rich in the aquifer. Magnesium ranged between 11 and 243 mg/l, with an average value of 43 mg/l. The presence of magnesium normally increased the

**Table 1** Correlation matrix for groundwater quality parameters (Pearson)

	EC	TDS	pH	K	Na	Ca	Mg	HCO <sub>3</sub>	Cl	SO <sub>4</sub>	Fe	Mn	Cd	Zn	Pb	Cr	Co	Cu	Al
EC	1.00																		
TDS	0.83	1.00																	
pH	-0.47	-0.62	1.00																
K	0.83	0.68	-0.26	1.00															
Na	0.96	0.87	-0.51	0.80	1.00														
Ca	0.87	0.59	-0.21	0.68	0.80	1.00													
Mg	0.88	0.82	-0.57	0.76	0.84	0.69	1.00												
HCO <sub>3</sub>	-0.40	-0.31	0.50	-0.29	-0.31	-0.17	-0.53	1.00											
Cl	0.94	0.87	-0.60	0.79	0.92	0.73	0.97	-0.55	1.00										
SO <sub>4</sub>	0.97	0.81	-0.43	0.89	0.93	0.79	0.88	-0.45	0.93	1.00									
Fe	-0.11	-0.22	0.31	-0.09	-0.12	0.02	-0.25	0.18	-0.19	-0.11	1.00								
Mn	0.14	0.00	0.13	0.09	0.06	0.17	0.04	-0.06	0.05	0.18	0.15	1.00							
Cd	-0.14	-0.10	0.17	0.02	-0.06	-0.15	-0.12	0.19	-0.12	-0.10	-0.01	-0.15	1.00						
Zn	-0.09	-0.25	0.15	-0.19	-0.13	0.08	-0.19	0.13	-0.20	-0.07	0.31	0.63	-0.20	1.00					
Pb	-0.20	-0.16	0.08	-0.16	-0.18	-0.13	-0.16	-0.02	-0.14	-0.17	0.83	-0.03	-0.07	0.12	1.00				
Cr	0.02	-0.26	0.10	-0.14	0.02	0.28	-0.22	0.16	-0.15	-0.10	0.17	-0.09	-0.04	0.14	-0.15	1.00			
Co	-0.24	-0.26	-0.08	0.01	-0.22	-0.29	-0.18	-0.16	-0.19	-0.19	-0.09	-0.25	0.28	-0.19	0.04	0.03	1.00		
Cu	-0.09	-0.11	0.04	-0.13	-0.09	-0.06	-0.11	0.04	-0.12	0.00	0.16	0.51	-0.09	0.77	0.02	0.08	-0.22	1.00	
Al	0.00	-0.06	0.08	0.10	-0.01	0.04	0.02	-0.08	0.02	0.10	0.89	0.24	0.00	0.45	0.86	-0.12	-0.08	0.44	1.00



**Table 2** Limits of WHO, Egyptian standards and assigned weight, and relative for chemical parameters

Parameters	WHO desirable limit (mg/l) (a)	WHO allowable limit (mg/l) (b)	Egypt limit (mg/l)	Weight (wi)	Relative weight (Wi)
				(wi)	(Wi)
EC	1000 (mmhos)	1500 (mmhos)	–	5	0.082
TDS	500	1000	1000	4	0.066
pH	6.5–8.5	8.5	6.5–8.5	2	0.033
K	10	12	12	5	0.082
Na	200	200	200	5	0.082
Ca	75	75	200	3	0.049
Mg	50	50	150	5	0.082
HCO <sub>3</sub>	250	500	–	2	0.033
co <sub>3</sub>	100	100	–	2	0.033
Cl	250	250	250	5	0.082
SO <sub>4</sub>	250	250	250	4	0.066
Fe	0.3	1	0.3	2	0.033
Mn	0.05	0.1	0.4	2	0.033
Cd	0.005	0.1	0.003	2	0.033
Zn	5	5	3	2	0.033
Pb	0.01	0.01	0.01	2	0.033
Cr	0.05	0.05	0.05	2	0.033
Co	0.05	0.05	1	2	0.033
Cu	1	2	2	2	0.033
Al	0.2	0.2	0.2	2	0.033
				$\sum w_i = 61$	$\sum W_i = 1$

alkalinity of the soil and groundwater [10, 37]. Calcium ranged between 12 and 714 mg/l with a mean value of 119 mg/l. For all the collected groundwater samples, calcium concentration is higher than magnesium. This can be explained by the abundance of carbonate minerals that compose the water-bearing formations as well as ion exchange processes and the precipitation of calcite in the aquifer. Chloride content for groundwater samples varies between 18 and 2662 mg/l with an average value of 423 mg/l. Carbonate was not detected in groundwater, while bicarbonate ranged from 85 to 500 mg/l. Figures 5, 6, and 7 were drawn to show the extent of variation between the samples in each well.

Piper diagram [30] was used to identify the groundwater type in the study area as shown in Fig. 4. According to the prevailing cations and anions in groundwater samples Na–Cl water type in the southern area due to salinity of the Miocene aquifer, Mg–HCO<sub>3</sub> water type in the northern area due to seepage from Ismailia Canal and excess of irrigation water and there is an interference zone which has a mixed water type between marine water from south and fresh water from north.

Atta, et al. [5] revealed that the abundance of Fe, Mn, and Zn in the groundwater is due to geogenic aspects, not pollution sources. Khalil et al. [26] and Awad et al. [6] revealed that the source of groundwater in the area is greatly affected by freshwater seepage from canals and excess irrigation water which all agreed with the study.

**Table 3** The permissible limits WHO and results of the physio-chemical and environmental parameters with aerial distribution of wells%

Parameter	WHO desirable limit (mg/l) (a)	WHO allowable limit (mg/l) (b)	Undesirable limit (mg/l) (c)	Min.	Max.	Avg.	Med.	%Wells according permissible limits for Each parameter		
								(a)	(b)	(c)
EC	1000 (mmhos)	1500 (mmhos)	> 1500 (mmhos)	0.34	9.00	2.03	0.98	100.0	0	0
TDS	500	1000	> 1000	263.0	5765	1276.69	646.5	43.79	29.16	27.05
pH	6.5–8.5	8.5	> 8.5	7.10	8.50	7.78	7.90	81.25	18.75	0
K	10	12	> 12	4.00	46.0	10.55	8.00	64.60	20.8	14.6
Na	200	200	> 200	31.00	1242	270.18	161.0	68.73	0	31.27
Ca	75	75	> 200	12.00	714.0	118.82	33.00	70.80	12.5	16.7
Mg	50	50	> 150	11.00	243.0	43.47	23.00	85.45	6.25	8.3
HCO <sub>3</sub>	250	500	> 500	85.00	500.0	246.06	219.0	66.70	33.3	0
Cl	250	250	> 250	18.00	2662	422.71	145.0	72.90	22.9	4.2
SO <sub>4</sub>	250	250	> 250	12.00	1108	184.1	52.00	79.20	0	20.8
Fe	0.3	1	> 0.3	0.00	1.09	0.13	0.02	86.40	13.6	0
Mn	0.05	0.1	> 0.1	0.01	0.78	0.21	0.10	27.30	22.7	50
Cd	0.005	0.1	> 0.1	0.00	0.02	0.00	0.00	77.40	9	13.6
Zn	5	5	> 5	0.00	0.73	0.16	0.11	100.0	0	0
Pb	0.01	0.01	> 0.05	0.00	0.64	0.06	0.03	27.20	36.4	36.4
Cr	0.05	0.05	> 0.05	0.00	0.04	0.01	0.01	100.0	0	0
Co	0.05	0.05	> 0.05	0.01	0.02	0.01	0.01	100.0	0	0
Al	0.2	0.2	> 0.2	0.01	0.37	0.05	0.01	96.0	0	4

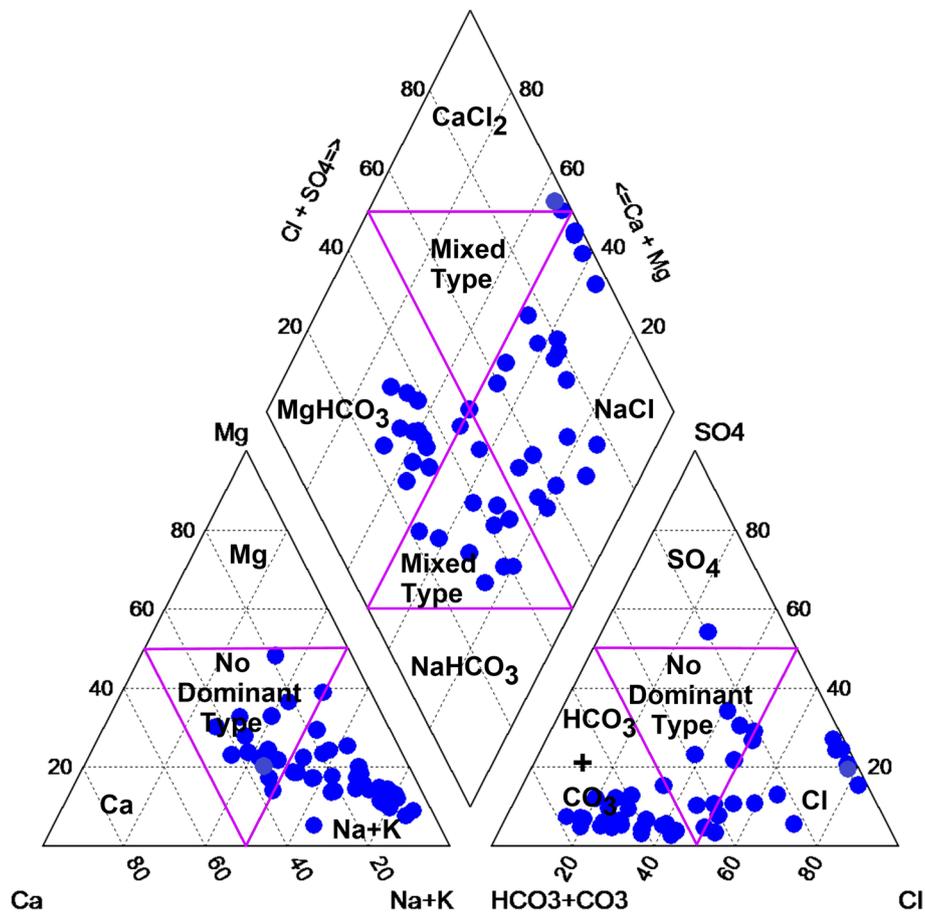


Fig. 4 Piper trilinear diagram for the groundwater samples

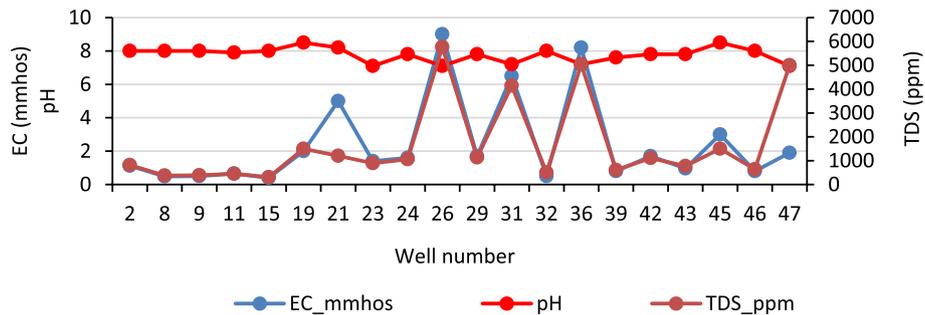


Fig. 5 Concentration of selected physio-chemical parameters

Table 3 and Fig. 8 showed that 100% of wells for EC were assigned at desirable limits. 43.79% of wells for TDS were assigned at the desirable limit and 27.05% of them at the undesirable limits. While pH, 81.25% were assigned at the desirable limit. The percentage of wells for the aerial distribution of cations concentration assigned at desirable limits ranged between 64.6% for K, 85.45% for Mg, 68.73% for Na, and 70.8% for Ca. While the percentage of wells for the aerial distribution of cations concentration

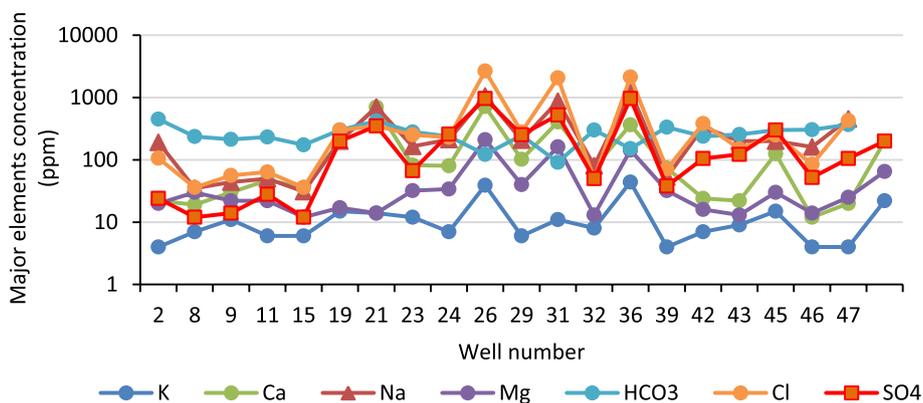


Fig. 6 Concentration of major elements

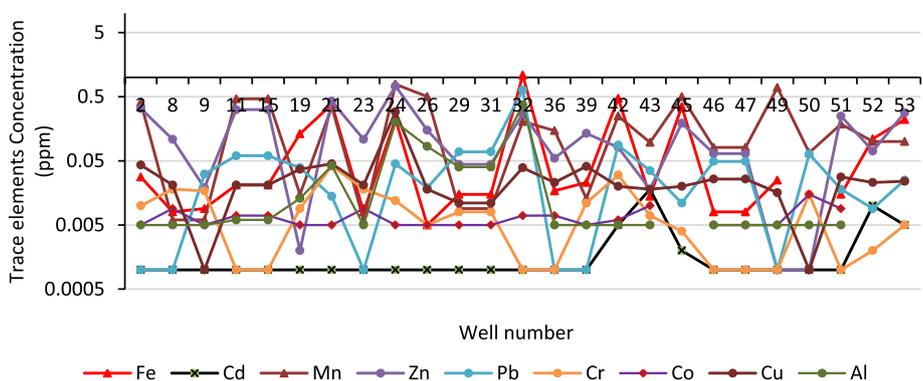


Fig. 7 Concentration of trace element

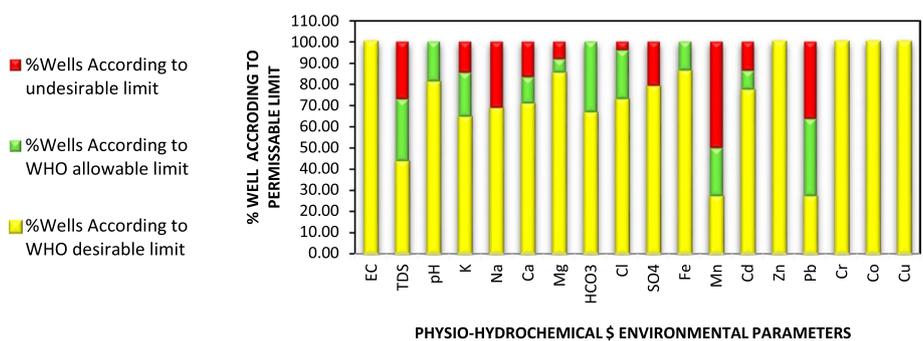


Fig. 8 Concentration for 20 elements by percentage of wells (relevant to their limits of WHO for each element)

assigned at the undesirable limits ranged between 8.3% for Mg, 31.27% for Na, 14.6% for K, and 16.7% for Ca.

The percentage of wells for the aerial distribution of anions concentration assigned at desirable limits ranged between 72.9% for Cl, 66.7% for HCO<sub>3</sub>, and 79.2% for SO<sub>4</sub>. While the percentage of wells for the aerial distribution of anions concentration assigned at the undesirable limit ranged between 4.2% for Cl, 0% for HCO<sub>3</sub>, and 20.8% for SO<sub>4</sub> as shown in Table 3 and Fig. 8.

Table 3 and Fig. 8 presented the aerial distribution concentration for 8 sensitive trace elements. The percentage of wells assigned at desirable limits ranged between 100% for (Zn, Cr, and Co), 86% for Fe, 27.3% for Mn, 77.4% for Cd, 27.2% for Pb, and 96% for Al, while the percentage of wells assigned at undesirable limits ranged between 0% for (Fe, Zn, Cr, and Co), 50% for Mn, 13.6% for Cd, 36.4% for Pb, and 4% for Al.

Figure 8 summarizes the results of the concentration for the selected 20 elements (11 physio-hydrochemical characteristics, and 9 sensitive environmental trace elements) by %wells relevant to the limits of WHO for each element.

The water quality index is one of the most important methods to observe groundwater pollution (Alam and Pathak, 2010) [1] which agreed with the results. It was calculated by using the compared different standard limits of drinking water quality recommended by WHO (2008) and Egyptian Standards (2007). Two values for WQI were calculated and drawn according to these two standards. It was classified into six classes relevant to the drinking groundwater quality classes: excellent water (WQI < 25 mg/l), good water (25–50 mg/l), poor water (50–75 mg/l), very poor water (75–100 mg/l), undesirable water (100–150 mg/l), and unfit water for drinking water (> 150 mg/l) as shown in Fig. 9a, b. Figure 9a (WHO classification) indicated that in the most parts of the study area, the good water class was dominant and reached to 35.8%, 28.8% was excellent water; 7.5% were poor water, 11.3% very poor water quality, and 13.3% were unfit water for drinking water. Similarly, for Egyptian Standard classification via WQI, the study area was divided into six classes: Fig. 9b indicated that 35.8% of groundwater was categorized as excellent water quality, 34% as good water quality, 9.4% as poor water, 5.7% as very poor water, 1.9% as undesirable water and 13.3% as unfit water quality. This assessment was compared to Embaby et al. [20], who used WQI in the assessment of groundwater quality in El-Salhia Plain East Nile Delta. The study showed that 70% of the analyzed groundwater samples fall in the good class, and the remainder (30%), which were situated in the middle of the plain, was a poor class which mostly agreed with the study.

### Conclusions and recommendation

This research studied the groundwater quality assessment for drinking using WQI and concluded that most of observation wells are located within desirable and max. allowable limits.

The groundwater in the study area is alkaline. TDS in groundwater ranged from 263 to 5765 mg/l, with a mean value of 1277 mg/l. Sodium and chloride are the main cation and anion constituents.

The water type is Na–Cl in the southern area due to salinity of the Miocene aquifer, Mg–HCO<sub>3</sub> water type in the northern area due to seepage from Ismailia Canal and excess of irrigation water and there is an interference zone which has a mixed water type between marine water from south and fresh water from north.

The WQI relevant to WHO limits indicated that 23% of wells were located in excellent water quality class that could be used for drinking, irrigation and industrial uses, 38% of wells were located in good water quality class that could be used for domestic, irrigation, and industrial uses, 11% of wells were located in poor water quality class that could be used for irrigation and industrial uses, 8% of wells were located in very poor

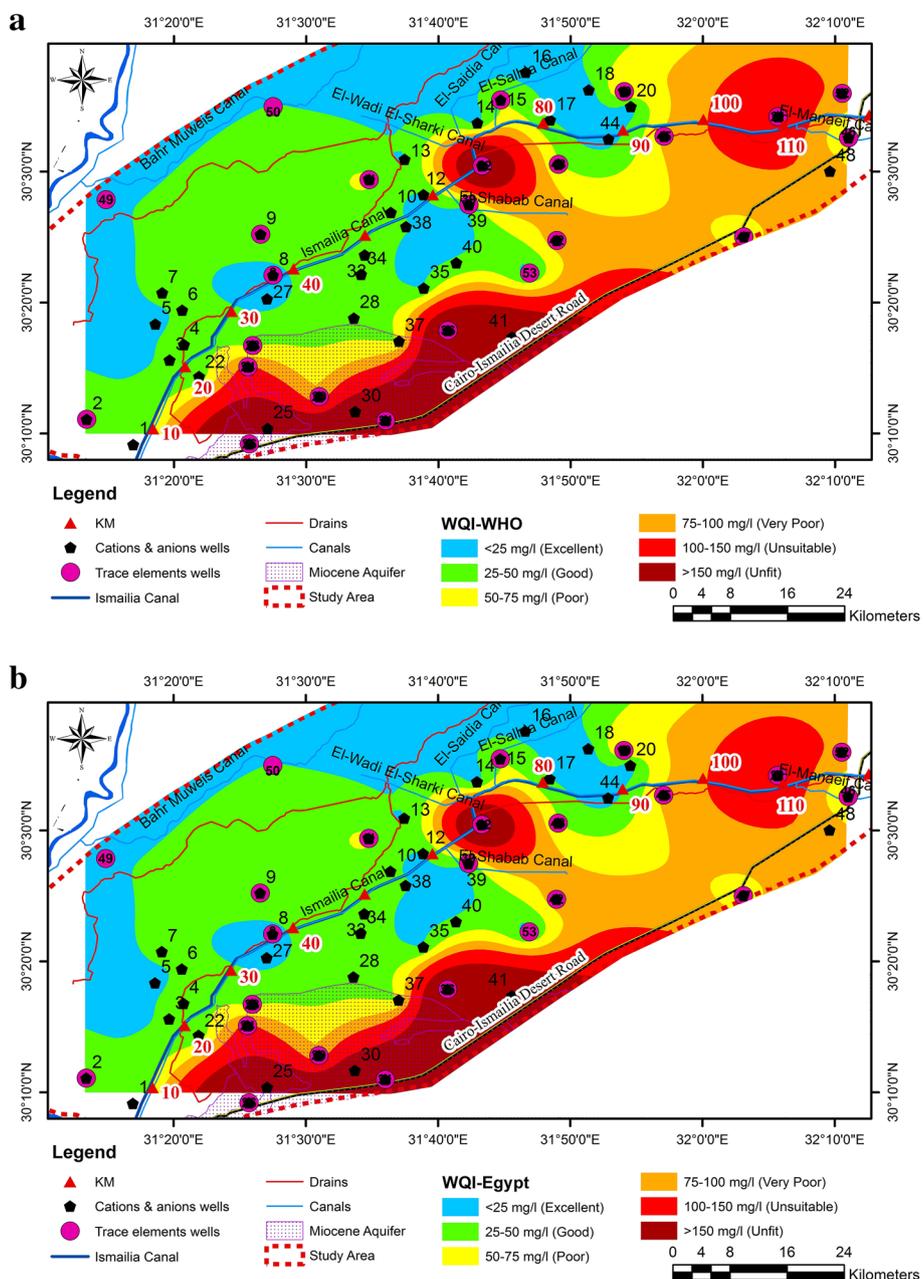


Fig. 9 a, b WQI aerial distribution for drinking groundwater suitability for WHO (a) and Egyptian standards (b)

water quality class that could be used for irrigation, 6% of wells were located in unsuitable water quality class which is restricted for irrigation use and 15% of wells were located in unfit water quality which will require proper treatment before use.

The WQI relevant to Egyptian standard limits indicated that 25% of wells were located in excellent water quality class that could be used for drinking, irrigation, and industrial uses, 43% of wells were located in good water quality class that could be used for domestic, irrigation, and industrial uses, 8% of wells were located in poor water quality class that could be used for irrigation and industrial uses, 6% of wells were located in

very poor water quality class that could be used in irrigation, 6% of wells were located in unsuitable water quality class which is restricted for irrigation use and 13% of wells were located in unfit water quality which will require proper treatment before use.

The percentage of wells located at unfit water for drinking were assigned in the Miocene aquifer, and north of Ismailia Canal between km 67 to km 73 and from km 95 to km 128.

It is highly recommended to study the water quality of the Ismailia Canal which may affect the groundwater quality. It is recommended to study the water quality in detail between km 67 to 73 and from km 95 to km 128 as the WQI is unfit in this region and needs more investigations in this region. A full environmental impact assessment should be applied for any future development projects to maximize and sustain the groundwater as a second resource under the area of Ismailia Canal.

#### Abbreviations

WHO	World Health Organization
WQI	Water Quality Index
Ec	Electrical conductivity
TDS	Total dissolved solids
Na	Sodium
K	Potassium
Ca	Calcium
Mg	Magnesium
Cl	Chloride
CO <sub>3</sub>	Carbonate
SO <sub>4</sub>	Sulphate
HCO <sub>3</sub>	Bicarbonate
Fe	Iron
Mn	Manganese
Zn	Zinc
Cu	Copper
Pb	Lead
Co	Cobalt
Cr	Chromium
Cd	Cadmium
Al	Aluminium

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#### Authors' contributions

HS: investigation, methodology, writing—original draft. MA: investigation, writing—original draft and reviewing. AM: reviewing and editing. The authors read and approved the final manuscript.

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#### Availability of data and materials

The datasets generated and analyzed during the current study are not publicly available because they are part of a PhD thesis and not finished yet but are available from the corresponding author on reasonable request.

#### Declarations

##### Ethics approval and consent to participate

Not applicable.

##### Consent for publication

Not applicable.

##### Competing interests

The authors declare that they have no competing interests.

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